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N-Boc-L-Valine-Connected Amidomonophosphane Rhodium(I) Catalyst for Asymmetric Arylation of *N*-Tosylarylimines with Arylboroxines

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The asymmetric synthesis of diarylmethylamines is very important because these amines are subunits of some biologically significant compounds.1 However, the efficient asymmetric synthesis of these amines is rather limited.²⁻⁴ Despite many reports, only two examples of the catalytic asymmetric additions of arylmetallic reagents to arylimine derivatives have been reported.5 Hayashi developed a chiral MOP-based phosphane rhodium(I)catalyzed addition of arylstannanes to N-sulfonylimines.⁶ A diphenylzinc addition to masked N-formylimines employing a paracyclophane-based ketimine catalyst was reported by Bräse.7 From the viewpoint of green chemistry, however, the catalytic asymmetric reaction of imines with less toxic arylation reagents is desirable.^{8,9} We selected arylboron reagents¹⁰ as an aryl source and succeeded in the development of the efficient asymmetric arylation of imines. The key points to the success were (1) L-valine-connected amidomonophosphane as a chiral ligand to rhodium(I), (2) steric tuning of arylimines, and (3) arylboroxines instead of arylboronic acids.

We started our arylation approach with hemilabile chiral amidomonophosphanes **1** and **2** (Figure 1), which were proven to be effective in two different types of asymmetric addition reactions. The less bulky phosphane **1**-rhodium(I) catalyzes the conjugate addition of arylboronic acids to enones,¹¹ and the more bulky **2**-copper(I) mediates the 1,2-addition of dialkylzinc reagents to alkyl- and arylimines.¹² Unfortunately, both phosphanes were ineffective in the asymmetric rhodium(I)-catalyzed arylation of 4-tolylaldehyde *N*-toluenesulfonylimine **8a** with phenylboronic acid **9** at 100 °C for 1 h in dioxane, giving *N*-tosyl-diarylmethylamide **10a** in marginal stereoselectivity (Scheme 1; Table 1, entries 1 and 2). BINAP was also not the good ligand, giving **10a** with 67:33 er (entry 3).

The replacement of the pivaloyl group of **1** and **2** by a readily available chiral α -amino acid is the advantage of the phosphane scaffold. Although *N*-Boc-D- and L-valine-connected bulky amidophosphanes **4** and **5** were less effective, giving **10a** with 53:47 er and 62:38 er in low chemical yields, less bulky **6** and **7** gave better reactivity (over 90% yield within 1 h) and selectivity (70:30 er and 76:24 er (entries 4–7)).

Encouraged by the enantioselectivity improvement with **7**, we then turned our attention to the steric tuning of the imine. Enantioselectivity was found to be dependent on the position of a substituent on the phenyl ring. Tolylaldehyde imines **8b** and **8c** bearing 3- and 2-methyl substituents were converted to **10b** and **10c** using 3 mol % of **7**-Rh(I) catalysis with better 78:22 er and 87:13 er (entries 8 and 9). Since a trimethylsilyl (TMS) group on a phenyl ring is easily convertible to a proton or halogen, the arylation was examined with 2-TMS-benzaldehyde¹³ imine **8d**. The enantioselectivity was improved to 89:11 er to give **10d** (entry 10).

The **7**-Rh(I)-catalyzed arylation of **8d** with phenylboronic acid **9** in dioxane, THF, and *n*-PrOH at 60 °C gave **10d** with the same 90:10 er in 8, 20, and 72% yields (entry 11). The chemical yield



Figure 1. Chiral phosphane ligands 1-7.

Scheme 1. Asymmetric Arylation of *N*-Tosylarylimine 8 with Phenylboron 9 and 11a, Giving 10



Table 1. Asymmetric Arylation of *N*-Tosylarylimine 8 with Phenylboron 9 and 11a Catalyzed by 1–7 and Rhodium(I)

| entry | imine 8 | R^4 | 1–7 | boron | solvent | yield (%) | era |
|-------|---------|-------|-----|-------|---------|-----------|-------|
| 1 | 8a | 4-Me | 1 | 9 | dioxane | 80 | 51:49 |
| 2 | 8a | 4-Me | 2 | 9 | dioxane | 20 | 53:47 |
| 3 | 8a | 4-Me | 3 | 9 | dioxane | 88 | 67:33 |
| 4 | 8a | 4-Me | 4 | 9 | dioxane | 26 | 53:47 |
| 5 | 8a | 4-Me | 5 | 9 | dioxane | 24 | 62:38 |
| 6 | 8a | 4-Me | 6 | 9 | dioxane | 91 | 70:30 |
| 7 | 8a | 4-Me | 7 | 9 | dioxane | 95 | 76:24 |
| 8 | 8b | 3-Me | 7 | 9 | dioxane | 99 | 78:22 |
| 9 | 8c | 2-Me | 7 | 9 | dioxane | 99 | 87:13 |
| 10 | 8d | 2-TMS | 7 | 9 | dioxane | 91 | 89:11 |
| 11 | 8d | 2-TMS | 7 | 9 | n-PrOH | 72 | 90:10 |
| 12 | 8d | 2-TMS | 7 | 11a | n-PrOH | 95 | 90:10 |

^a Determined by HPLC (Supporting Information).

of **10d** with 90:10 er was improved to 95% by using phenylboroxine $((PhBO)_3)$ **11a** in *n*-PrOH at 60 °C for 3 h (entry 12).

It was an unexpected pleasure to find that the arylation of **8d** with substituted phenylboroxines **11b**–**11f** gave **10** with higher enantioselectivity up to 97:3 (Scheme 2). For example, 3-chlorophenylation of **8d** with **11f** gave **10df**¹⁴ with 97:3 er quantitatively (Table 2, entry 9). The electron-donating 3-methoxyphenyl group was introduced to **8d** with **11e**, giving **10de** with 95:5 er in 87% yield (entry 8). 4-Methoxy- and 4-chlorophenyl groups were also introduced to **8d** with **11c** and **11d**, giving **10dc** and **10dd** with 94:6 er (84%) and 95:5 er (97%) (entries 6 and 7). 4-Phenylphenylboroxine **11b** was also a good donor to give **10db** with 96:4 er in 98% yield (entry 4).

Other than 2-TMS-phenylimine **8d**, 2-, 3-, and 4-tolylimines **8a**–**8c** were converted, upon treatment with 4-phenylphenylboroxine



| Ar1 NITS + | (Ar ² BO) ₃ | 7-Rh(acac)(C ₂ H ₄) ₂ 3 mol % | Ar ² |
|---|--|---|-----------------------------------|
| 8 a : Ar ¹ = 4-MeC ₆ H ₄ b : Ar ¹ = 3-MeC ₆ H ₄ | 1.7 eq 11 a: Ar ² = Ph b: Ar ² = 4-P | <i>n</i> -PrOH 60-100 °C, 1-3 h 'hC ₆ H₄ | Ar ¹ N/10 10 |
| c : Ar ¹ = 2-MeC ₆ H ₄ d : Ar ¹ = 2-TMSC ₆ H ₄ e : Ar ¹ = Ph f : Ar ¹ = 1-Naphthyl | c : $Ar^2 = 4-N$ d : $Ar^2 = 4-C$ e : $Ar^2 = 3-N$ f : $Ar^2 = 3-C$ | 1eOC ₆ H ₄ CIC ₆ H ₄ 1eOC ₆ H ₄ IC ₆ H ₄ | |

Table 2. Rhodium(I)-7-Catalyzed Asymmetric Arylation of *N*-Tosylarylimines **8** with Arylboroxines **11** in *n*-PrOH, Giving **10**

| | | | | | temp | yield | |
|-------|----|------------------------------------|-----|------------------------------------|------|-------|-----------------|
| entry | 8 | Ar ¹ | 11 | Ar ² | (°C) | (%) | er ^a |
| 1 | 8a | 4-MeC ₆ H ₄ | 11b | 4-PhC ₆ H ₄ | 60 | 86 | 86:14 |
| 2 | 8b | 3-MeC ₆ H ₄ | 11b | 4-PhC ₆ H ₄ | 60 | 90 | 88:12 |
| 3 | 8c | 2-MeC ₆ H ₄ | 11b | $4-PhC_6H_4$ | 60 | 97 | 93:7 |
| 4 | 8d | 2-TMSC ₆ H ₄ | 11b | $4-PhC_6H_4$ | 60 | 98 | 96:4 |
| 5 | 8e | Ph | 11b | $4-PhC_6H_4$ | 60 | 83 | 83:17 |
| 6 | 8d | 2-TMSC ₆ H ₄ | 11c | 4-MeOC ₆ H ₄ | 80 | 84 | 94:6 |
| 7 | 8d | 2-TMSC ₆ H ₄ | 11d | 4-ClC ₆ H ₄ | 80 | 97 | 95:5 |
| 8 | 8d | 2-TMSC ₆ H ₄ | 11e | 3-MeOC ₆ H ₄ | 60 | 87 | 95:5 |
| 9 | 8d | 2-TMSC ₆ H ₄ | 11f | 3-ClC ₆ H ₄ | 60 | 99 | 97:3 |
| 10 | 8f | 1-Naphthyl | 11b | $4-PhC_6H_4$ | 100 | 88 | 96:4 |

^a Determined by HPLC or NMR (Supporting Information).





11b, to the corresponding diarylmethylamides **10** with 93:7 er, 88: 12 er, and 86:14 er (entries 1–3). Naphthylimine **8f** was also a good acceptor with **11b**, being converted to **10fb** with 96:4 er in 88% yield (entry 10).

The 2-TMS group on the phenyl ring of the product was easily convertible to other functional groups (Scheme 3). Thus, **10db** was protodesilylated with cesium fluoride in refluxing aqueous DMF for 25 h to afford (*R*)-**10eb**¹⁵ in 74% yield without any racemization. This arylation of the TMS-modified imine-protodesilylation sequence is complementary to the less effective arylation of phenylimine **8e** (Table 2, entries 4 vs 5). Iododesilylation of **10dd** with iodine chloride in methylene chloride at 0 °C for 15 min gave 2-iodo derivative **12**, a useful starting material suitable for the Suzuki–Miyaura coupling, quantitatively without any racemization. Treatment of **10eb** with samarium iodide in refluxing THF–HMPA for 6 h gave the corresponding detosylated (*R*)-**13**¹⁶ in 98% yield without any loss of optical purity. The *si*-face attack to **8** in the arylation was determined from the established absolute configuration of **10eb** and **13**.¹⁶

In conclusion, a catalytic asymmetric arylation of sterically tuned imines with arylboroxines was developed by using *N*-Boc-L-valine-connected amidomonophosphane rhodium(I) catalyst in *n*-PrOH. It is also important to note that further modification of the amidophosphane is possible with use of other natural α -amino acids. The TMS group used for the steric tuning of the imine is convertible to other functionalities that are applicable as a key foothold for the carbon–carbon bond forming coupling reactions.

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Supporting Information Available: Experimental procedure, characterization data, NMR spectra, and HPLC traces (PDF). This material is available free of charge via the Internet at http://pubs.acs.org.

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